Analysis of polarization degree of monochromatic line in H₂-Ne gas mixture

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The purpose of this paper is to analyse the variance of the Polarization Degree (PD) for the mixture of gases H₂-Ne, using the intensity of the monochromatic line (λ =585,23nm) measuring at different values of the discharge current for different values of the pressure. Such a research can be integrated in the statistical model 'Analysis of Variance (ANOVA)', which offers the possibility to follow the evolution of a phenomenon under different conditions. Said in another way, we expect that the means obtained for the four values of the pressure to differ enough in order to decide that their variance is connected with the cumulative effect of the polarization degree, induced by the different values of the concentration of the mixture.

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Keywords: Polarization Degree (PD), Monochromatization-effect (M-effect), Analysis Of Variance (ANOVA) method

1. Introduction

In the last 20 years, studies performed in electronegative-electropositive gas mixtures plasma, at a total pressure varying in the range of ten to hundred mbar, reported the appearance of a quasi-mochromatization effect of the emitted radiation. Due to the fact that the modified spectrum is reduced to this dominant line, the phenomenon was called the "Monochromatisation-effect" (M-effect).

For instance, in Ne+H₂ gas mixtures plasma in a dielectric barrier discharge (DBD), the M-effect consists in the reduction of the discharge emission spectrum practice at one strong spectral line $\lambda = 585.3$ nm [1÷4]. Figures 1 show the emission spectrum of a dielectric barrier discharge in Ne+H₂ gas mixtures.



Fig. 1. Emission spectrum of DBD in $Ne+H_2$ gas mixtures.

The M-effect has proven to have a general character because its appearance was observed not only in A.C. but also in D.C. discharges, the main condition being that the gas mixture has to be formed from electronegative/electropositive gases, at moderate to high total pressures [5, 6].

The main reaction mechanism of the M-effect is represented by the resonant three-body polar reaction of ionized, excited and ground state atoms, with the important contribution of the electronegative gas metastable $[7\div10]$.

The general form of the reaction, in case of the M-effect, is:

$$P^{+} + N^{-} + N^{met} \rightarrow P^{*} + N^{graundstate} + \left(N^{met}\right)^{*} + \Delta E \left(\Delta E \approx 0\right)$$
(1)

where

P, N - the symbols of the atoms of electropositive and respectively, electronegative gases in the mixture,

 \mathbf{P}^+ - the symbol for the positive ion,

 N^{-} - the symbol of the negative ion,

 N^{met} – the symbol for the metastable negative atom, $(N^{met})^*$ - is the symbol of the excited electronegative atom standing in an upper state energy that the metastable level, P^* - the electropositive atom in an excited state, ΔE - the reaction energy defect.

This reaction is based on the existence of a third particle, in a convenient energetic state, namely the metastable atoms of the electronegative gas. Only if this condition is accomplished, the reaction becomes resonant:

$$Ne^{+} + H^{-} + H^{*}(n = 2) \rightarrow (Ne^{+}H^{-}) + H^{*}(n = 2) \rightarrow Ne^{*}(2p1) + H^{*}(n = 3) + H(n = 1) \pm \Delta E$$
(2)

For the gas mixtures in which the electronegative gas has a strong electronic affinity, like the chlorine atoms, the generation reaction for the M-effect could be binary in the classical sense of the Landau-Zenner theory [11-14], [17-20].

2. Experimental

In order to allow the passage of the UV radiation, the RF discharge was produced in a quartz tube with 16 mm inside diameter and 20 mm outside diameter respectively, between two identical wolfram-thorium cylinders electrodes of 12 mm diameter, spaced at 6 mm distance.

The experimental discharge device can be pumped down up to a vacuum value of 10^{-4} torr and then filled with various gas mixtures of spectral purity. The electrical power supply used in the experiment had the following characteristics: maximum output electrical tension of 2kV corresponding to a maximum electrical current intensity of 150mA, one optional frequency of 25 kHz. The optical emission spectra of the plasma discharges were registered using an Optical Analyzer Multichannel, spectral range 220÷900nm, integration time of 15ms and a resolution of 1.5nm. A photo - view of the experimental set - up is presented in Fig. 2. The measurements of the polarization degree were performed in hydrogen-neon gas mixtures $(v_{H2}/v_{Ne}=19/56 \text{ and } v_{H2}/v_{Ne}=1)$ for the dominant spectral line with λ =585.3nm, at four values of total pressures namely 19torr, 30torr, 45torr and 63torr. Each set of measurements was done for one frequency, at 25 kHz. The discharge electrical current intensity was varying with a rate of 2.5mA within the range of 6÷24mA.



Fig. 2. View of the experimental device.

The experimental data are presented in Fig. 3 and Fig. 4.



Fig. 3. Polarization Degree for $v_{H2}/v_{Ne}=19/56$.



Fig. 4. Polarization Degree for $v_{H2}/v_{Ne}=1$.

3. Results and discussion

For the statistical analyses in this paper we used the IBM-SPSS Trial program, version 22.0.0 [15].

The research model supposes the variance analysis of a variable measured repeatedly in different conditions. Such a research can be integrated in the statistical model 'Analysis of Variance (ANOVA)' [16], which offers the possibility to follow the evolution of a phenomenon under different conditions. In such an analysis, the total variance (SS_T- the variance of the individual values of all the researched samples, independent of the group they belong, multiplied with n-1 degrees of freedom) of the values in the table below is fully provided by the variance withinsubject (SS_w-the within-subjects variance), which, in its turn, is composed of the variance provided by the levels of the independent variables (SS_M - the variance of the means of each group - condition of measurement compared with the total mean considering all values together) and the variance of the error, which is not explained by the independent variables $(SS_E - the sum of$ the variances for each condition of measurement). The F ratio describes the model in the statistical test,

$$F = \frac{SS_M}{SS_F}$$

The higher the variance induced by conditions of measurement (SS_M) compared to the variance of the

unexplained error (SS_E) , the higher the F value, being able in this way to reach the threshold of statistical significance.

Table 1. The PD Values for four values of the pressure

Current	PD_19torr	PD_30torr	PD_45torr	PD_63torr	average
5	0,1452529	0,1720359	0,171696	0,247684	0,184167
6	0,1313081	0,1819724	0,197573	0,243683	0,188634
7	0,1424355	0,1781508	0,208662	0,268158	0,199352
8	0,1364826	0,1686948	0,199517	0,262731	0,191856
9	0,1333516	0,1767493	0,21528	0,28227	0,201913
10	0,1221737	0,1867289	0,20045	0,251496	0,190212
11	0,1400368	0,1771772	0,209034	0,279885	0,201533
12	0,096526	0,1668905	0,194839	0,278316	0,184143
13	0,1370501	0,175391	0,218618	0,318635	0,212423
14	0,1231812	0,1644513	0,19084	0,279879	0,189588
15	0,1443839	0,175212	0,202756	0,306675	0,207257
16	0,1191825	0,1665651	0,198894	0,283145	0,191947
17	0,1337276	0,1430132	0,224452	0,30717	0,202091
18	0,1221166	0,1690128	0,203663	0,280084	0,193719
19	0,1304734	0,1568098	0,218	0,305458	0,202685
20	0,1206695	0,1632515	0,192693	0,298318	0,193733
21	0,1325371	0,1541711	0,205579	0,315454	0,201935
22	0,1254084	0,1645372	0,194551	0,305637	0,197533
average	0,1297943	0,1689342	0,202617	0,284149	0,196373

ANOVA one-way

In this study it has been analyzed the variance of the Polarization Degree (PD), $(I_{max}+I_{min})/(I_{max}-I_{min})$, for the mixture of gases H₂-Ne, $(v_{H2}/v_{Ne}=19/56)$, using the intensities of the monochromatic line (λ =585,23nm), I_{max} and I_{min} , measuring at different values of the discharge current, I [mA] at the frequency of 25 kHz for different values of the pressure (63torr, 45torr, 30torr and 19torr).

Measure: PolarisationDegree						
Pressure	Dependent Variable					
1	PD_19torr					
2	PD_30torr					
3	PD_45torr					
4	PD_63torr					

Table 2: Within-Subjects Factors

The Table 3 contains the means and the standard deviations for the PD and the four values of the gas mixture.

Table 3. Descriptive Statistics

	Mean	Std.	Ν
		Deviation	
PD_19torr	,12979430	,011719654	18
PD_30torr	,16893415	,010598993	18
PD_45torr	,20261650	,012342662	18
PD_63torr	,28414882	,023215864	18

The data in the Table 4 indicates a statistically significant variance (p<0.001) of the PD depending on the pressure, with a very high level of the effect size (0,980) and an observed maximum power (1,000).

1 u u e +. muu vu ue resis	Table 4:	Multivariate	Tests ^a
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Ef	fect	Value	F	Hypoth	Error	Sig.	Partial Eta	Noncent.	Observed	
				esis df	df	-	Squared	Parameter	Power	
	Pillai's	0.98	243 949 ^b	3 00	15.0	00	980	731 85	1 000	
	Trace	0,20	213,212	5,00	10,0	,00	,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,	/51,05	1,000	
	Wilks'	0.02	2/13 0/10 ^t	3.00	15.0	00	980	731.85	1.000	
Pressure	Lambda	0,02	273,777	5,00	15,0	,00	,700	751,05	1,000	
	Hotelling's	18 70	242 040 ^t	2 00	15.0	00	080	721.05	1.000	
	Trace	48,79	243,949	5,00	15,0	,00	,980	751,85	1,000	
	Roy's									
	Largest	48,79	243,949 ^t	3,00	15,0	,00	,980	731,85	1,000	
	Root									
a. Design: Intercept Within Subjects Design: Pressure										
b. Exact s	b. Exact statistic									
c. Compu	c. Computed using $alpha = .05$									

The same result we obtained using the 'Tests of Within-Subjects Effects' - see the table below. Analyzing the results on the line of the 'Sphericity Assumed' it can be observed that the F test ist statistically significant (p<0.0005), which confirms the

research hypothesis, according to which the value of the PD varies with the pressure. The dimension of the effect very high (0,952), and the observed power is maximum (1,000).

	Measure: Polarisation Degree								
Source		Sum of Squares	df	Mean Square	F	Sig.	Partial Eta Squared	Noncent. Parameter	Observed Power ^a
	Sphericity Assumed	,233	3	,078	334,6	,00	,952	1004	1,000
Pressure	Greenhouse- Geisser	,233	1,681	,138	334,6	,00	,952	562,5	1,000
	Huynh-Feldt	,233	1,844	,126	334,6	,00	,952	617,2	1,000
	Lower-bound	,233	1,000	,233	334,6	,00	,952	334,6	1,000
	Sphericity Assumed	,012	51	,000					
Error Pressure	Greenhouse- Geisser	,012	28,6	,000					
	Huynh-Feldt	,012	31,4	,000					
	Lower-bound	,012	17,0	,001					
a. Computed using alpha = ,05									

Table 5: Tests of Within-Subjects Effects

The Table of "Pairwise Comparisons' presents the significance of the differences between all the four mean pairs

of the PD for the values of the pressure. Statistically significant differences can be observed for all the values of the pressure.

Pairwise Comparisons										
Measure: PolarisationDegree										
(I) Pressure	(J)	Mean Difference (I-	Std. Error	Sig. ^b	95% Confidence Interval for Difference ^b					
(1) 11005011	Pressure	J)			Lower Bound	Upper Bound				
	2	-,039*	,003	,000	-,049	-,029				
1	3	-,073*	,004	,000	-,084	-,062				
	4	-,154*	,006	,000	-,173	-,136				
	1	,039*	,003	,000	,029	,049				
2	3	-,034*	,004	,000	-,046	-,021				
	4	-,115*	,007	,000	-,137	-,094				
	1	,073*	,004	,000	,062	,084				
3	2	,034*	,004	,000	,021	,046				
	4	-,082*	,005	,000	-,095	-,068				
	1	,154 [*]	,006	,000	,136	,173				
4	2	,115*	,007	,000	,094	,137				
	3	,082*	,005	,000	,068	,095				
Based on es	stimated ma	arginal means								
*. The mean	n difference	e is significant	at the ,0	5 level.						
b Adjustme	ent for mult	tiple compariso	ns Bor	ferroni						

Table 6. Tests of Within-Subjects Effects

The graphic below shows the PD mean variance for the four values of the pressure.



Fig. 5. The variance of PD depending on pressure.

As it can be observed the data sustain the hypothesis and the image illustrates the existence of a statistically significant global variation, and also significant differences from one pressure to another, with the highest increase of pressure from pressure 3, of 45torr, to pressure 4, of 63torr.

ANOVA two-way

For analyzing the variance with repeated measurements of type ANOVA two-way, the goal of the research is the evaluation of the polarization degree, considering the four different pressures for two values of the concentrations of the gas mixture. Said in another way, we expect that the means obtained for the four values of the pressure to differ enough in order to decide that their variance is connected with the cumulative effect of the polarization degree, induced by the different values of the concentration of the mixture.

Such a research can be considered a longitudinal, within-subjects research model, which offers the possibility to analyze the evolution of the Polarization Degree (PD) for different values of the concentration of the mixture.

Table 7. Within-Subjects Factors

Measure: PolarisationDegree						
Pressure	Mix	Dependent Variable				
1	1	PD_mix1_19torr				
1	2	PD_mix2_19torr				
r	1	PD_mix1_30torr				
2	2	PD_mix2_30torr				
3	1	PD_mix1_45torr				
5	2	PD_mix2_45torr				
4	1	PD_mix1_63torr				
4	2	PD_mix2_63torr				

The Table Descriptive Statistics shows the means and the standard deviations of the PD, for the four values of the pressure and the two concentration values of the gas mixture.

Table 8. Descriptive Statistics

	Mean	Std. Deviation	Ν
PD_mix1_19torr	,12979430	,011719654	18
PD_mix2_19torr	,13974417	,011118903	18
PD_mix1_30torr	,16893415	,010598993	18
PD_mix2_30torr	,18317847	,015286590	18
PD_mix1_45torr	,20261650	,012342662	18
PD_mix2_45torr	,18335876	,012830105	18
PD_mix1_63torr	,28414882	,023215864	18
PD_mix2_63torr	,18371767	,012953928	18

The Table Multivariate Tests shows a statistically significant variation of the PD (p<0.01), regarding the two factors (pressure and concentration of the mixture), with a high level of the size of the effect (0,936 - 0,982) and a maximum observed power (1,000).

Eff	ect	Value	F	Hypothesis	Error	Sig.	Partial Eta	Noncent.	Observed	
				df	df		Squared	Parameter	Power ^c	
e	Pillai's Trace	,982	278,505 ^b	3,00	15,0	,00	,982	835,514	1,000	
sur	Wilks' Lambda	,018	278,505 ^b	3,00	15,0	,00	,982	835,514	1,000	
res	Hotelling's Trace	55,7	278,505 ^b	3,00	15,0	,00	,982	835,514	1,000	
d	Roy's Largest Root	55,7	278,505 ^b	3,00	15,0	,00	,982	835,514	1,000	
	Pillai's Trace	,936	247,002 ^b	1,00	17,0	,00	,936	247,002	1,000	
	Wilks' Lambda	,064	247,002 ^b	1,00	17,0	,00	,936	247,002	1,000	
ix	Hotelling's Trace	14,5	247,002 ^b	1,00	17,0	,00	,936	247,002	1,000	
Μ	Roy's Largest Root	14,5	247,002 ^b	1,00	17,0	,00	,936	247,002	1,000	
*	Pillai's Trace	,955	106,990 ^b	3,00	15,0	,00	,955	320,971	1,000	
ure iv	Wilks' Lambda	,045	106,990 ^b	3,00	15,0	,00	,955	320,971	1,000	
ess M	Hotelling's Trace	21,4	106,990 ^b	3,00	15,0	,00	,955	320,971	1,000	
Ы	Roy's Largest Root	21,4	106,990 ^b	3,00	15,0	,00	,955	320,971	1,000	
	a. Design: Intercept									
	Within Subjects Design: Pressure + Mix + Pressure * Mix									
	b. Exact statistic									
		с.	Computed	using alpha =	,05					

Table 9. Multivariate Tests^a

If we use the univariate tests, we have to analyze first the Mauchly Test for the sphericity, which is significant not only for the pressure and for the mixture, but also for the interaction pressure*mixture.

Measure: PolarisationDegree										
Within	Mauchly's W	Approx.	df	Sig.	Epsilon ^b					
Subjects Effect		Chi-Square			Greenhouse-	Huynh-	Lower-			
					Geisser	Feldt	bound			
Pressure	,286	19,659	5	,001	,672	,763	,333			
Mix	1,000	,000	0		1,000	1,000	1,000			
Pressure * Mix	,375	15,408	5	,009	,671 ,762 ,7					
Tests the null hy	Tests the null hypothesis that the error covariance matrix of the orthonormalized transformed									
dependent variables is proportional to an identity matrix.										
a. Design: Intercept										
Within Subjects Design: Pressure + Mix + Pressure * Mix										
b. May be used t	to adjust the degre	es of freedom	for the a	veraged	l tests of signific	cance. Co	orrected			
tests are displaye	ed in the Tests of	Within-Subjec	ts Effects	s table.						

As the Mauchly test is significant for the three categories we use the Greenhouse-Geisser correction. As it is shown in the table below, both the main effects of the two factors and their interaction is statistically significant.

This means that the values of the PD vary by the pressure, by the concentration of the gas mixture and also by the effect of the interaction of the two factors.

Table 11.	Tests of Within-Subjects Effects	

Measure	: PolarisationDegree								
Source		Sum of	df	Mean	F	Sig	Partial Eta	Noncent.	Observed
		Squares		Square		•	Squared	Parameter	Power ^a
ė	Sphericity Assumed	,182	3	,061	321,85	,0	,950	965,5	1,00
Pressur	Greenhouse-Geisser	,182	2,02	,090	321,86	,0	,950	648,7	1,00
	Huynh-Feldt	,182	2,29	,080,	321,86	,0	,950	736,2	1,00
	Lower-bound	,182	1,00	,182	321,86	,0	,950	321,8	1,00
	Sphericity Assumed	,010	51	,000					
Error	Greenhouse-Geisser	,010	34,3	,000					
Pressure	Huynh-Feldt	,010	38,9	,000					
	Lower-bound	,010	17,0	,001					
Mix	Sphericity Assumed	,021	1	,021	247,00	,0	,936	247,0	1,00
	Greenhouse-Geisser	,021	1,00	,021	247,02	,0	,936	247,0	1,00
	Huynh-Feldt	,021	1,00	,021	247,02	,0	,936	247,0	1,00
	Lower-bound	,021	1,00	,021	247,02	,0	,936	247,0	1,00
ix)	Sphericity Assumed	,001	17	8,307E-					
				005					
	Greenhouse Geisser	,001	17,0	8,307E-					
Ę	Greenhouse-Geisser			005					
ror	Huynh-Feldt	,001	17,0	8,307E-					
Ēr				005					
	Lower-bound	,001	17,0	8,307E-					
	Lower bound			005					
Pressure * Mix	Sphericity Assumed	,076	3	,025	113,69	,0	,870	340,8	1,00
	Greenhouse-Geisser	,076	2,01	,038	113,69	,0	,870	228,8	1,00
	Huynh-Feldt	,076	2,29	,033	113,69	,0	,870	259,6	1,00
	Lower-bound	,076	1,00	,076	113,69	,0	,870	113,6	1,00
Error(Pre ssure*Mi x)	Sphericity Assumed	,011	51	,000					
	Greenhouse-Geisser	,011	34,2	,000					
	Huynh-Feldt	,011	38,8	,000					
	Lower-bound	,011	17,0	,001					
o Comp	uted using elpha - 05								

a. Computed using alpha = .05

The post-hoc test of the multiple comparisons for the two factors shows the existence of a statistically

significant difference between all the values of the pressure.

Measure: PolarisationDegree							
(I)	(J)	Mean	Std.	Sig. ^b	95% Confidence Interval for		
Pressure	Pressure	Difference (I-J)	Error		Difference ^b		
					Lower Bound	Upper Bound	
1	2	-,041*	,003	,000	-,049	-,034	
	3	-,058*	,003	,000	-,068	-,048	
	4	-,099*	,003	,000	-,110	-,089	
2	1	,041*	,003	,000	,034	,049	
	3	-,017*	,002	,000	-,022	-,012	
	4	-,058*	,004	,000	-,070	-,046	
3	1	$,058^{*}$,003	,000	,048	,068	
	2	,017*	,002	,000	,012	,022	
	4	-,041*	,004	,000	-,052	-,029	
4	1	,099*	,003	,000	,089	,110	
	2	$,058^{*}$,004	,000	,046	,070	
	3	,041*	,004	,000	,029	,052	
Based on estimated marginal means							
*. The mean difference is significant at the ,05 level.							
b. Adjustment for multiple comparisons: Bonferroni.							

Table 12. Pairwise Comparisons

The picture below illustrates the variation of the Marginal Means of PD for the two factors, at the four values of the pressure.

For the first concentration of the gas mixture, the Marginals Means of PD increases progressively from one pressure to another. For the second concentration of the gas mixture ($v_{H2}/v_{Ne} = 1$) it can be observed that starting with pressure 2 (30 torr) the Marginal Means of PD become stationary.



Fig. 6. The variance of PD depending on pressure for tow concentration of gas mixture.

4. Conclusion

The goal of this paper was the statistical analysis of Polarization Degree of Monochromatic Line in H_2 -Ne Gas Mixture. First it has been analyzed the variance of the

Polarization Degree for the mixture of gases $H_2\text{-}Ne,\,\nu_{H2}\!/\!\nu_{Ne}$ =19/56, for different values of the pressure (63torr, 45torr, 30torr and 19torr) using ANOVA method. We obtained a statistically significant global variation for PD, and also significant differences from one pressure to another, with the highest increase of pressure from pressure of 45 torr, to pressure of 63 torr. Then we analyzed the evaluation of the polarization degree, considering the four different pressures for two values of the concentrations of the gas mixture. For the first concentration, $v_{H2}/v_{Ne} = 19/56$, of the gas mixture, the global variation of PD increases progressively from one pressure to another. For the second concentration of the gas mixture ($v_{H2}/v_{Ne} = 1$), starting with pressure of 30 torr the global variation of PD become stationary. As we expected, the means obtained for the four values of the pressure differ enough in order to decide that their variance is connected with the cumulative effect of the polarization degree, induced by the different values of the concentration of the mixture.

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