

XPS study of RNi_4B compounds, where $R = Nd, Tb, Dy, Ho$ and Er

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The RNi_4B compounds with $R = Nd, Tb, Dy, Ho$ and Er crystallize in $CeCo_4B$ -type structure. XPS measurements were performed at room temperature. The $Ni\ 2p_{1/2}$ and $Ni\ 2p_{3/2}$ core level lines reproduce well those of pure nickel. In all samples, the presence of $Ni\ 6\ eV$ satellite line was shown. This indicates the presence of holes in $Ni\ 3d$ band, in agreement with magnetic measurements. The $R\ 4d$ core levels are similar to those evidenced in pure R metals. The $4f$ orbitals keep their localized character and therefore the XPS spectra show multiplet structures. The small energy shift of $R4d$ and $R4f$ core level lines, as compared to R metals, may be correlated with minor changes in shielding which result from modification of electron concentrations when compounds are formed.

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1. Introduction

The RNi_4B compounds, where R is a rare earth, crystallize in a $CeCo_4B$ type structure, having $P6/mmm$ space group [1, 2]-Fig.1. The rare earth atoms occupy two nonequivalent positions: $1a$ and $1b$, the Ni atoms are situated in $2c$ and $6i$ positions and the B atoms in $2d$ position. The magnetic properties of these systems were analysed [3-5]. The XPS measurements on RNi_4B with $R = Ce, Pr$ and Nd [6] showed the presence of hybridization effects. The magnetic moments at $0\ K$ was shown to be very weak and dependent on the rare-earth partner. In the paramagnetic range an effective magnetic moment of $1.8\ \mu_B/Ni$ atom was shown [5].

As an on going work on the matter we report in this paper the results of XPS measurements on RNi_4B compounds with Tb, Dy, Ho, Er and Nd .

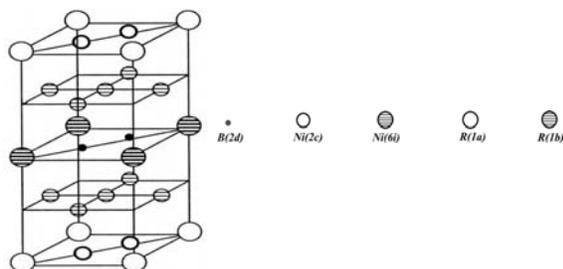


Fig. 1. The $CeCo_4B$ -type structure.

2. Experimental

The RNi_4B compounds were prepared by arc-melting the high purity elements, under argon atmosphere. The samples were remelted several times to ensure good homogeneity and then were thermally treated at $850\ ^\circ C$. The X-Ray analyses show the presence of a single phase.

The compounds crystallize in a $CeCo_4B$ – type structure.

XPS measurements were performed using a PHI 5600 MultiTechnique System. All spectra were recorded at room temperature, using monochromatized $Al\ K_\alpha$ radiation ($1486.6\ eV$). The total resolution, as determined at the Fermi level of a gold foil, was about $0.3-0.4\ eV$. The binding energies are given with reference to the Fermi level. The $4f_{7/2}$ gold level was found at $84.0\ eV$ binding energy. All spectra were recorded in vacuum below 5×10^{-10} mbar.

Band structure calculations were performed using the LMTO-ASA method [7,8].

3. Experimental results

Band structure calculations were performed for compounds with Gd, Tb, Dy, Ho and Er . The total densities of states are represented in Fig.2, for compounds with $R = Gd, Dy, Er$. The Ni atoms have a very small magnetic moment, M_{Ni} , at $0\ K$. The M_{Ni} values increase linearly with the De Gennes factor from $0.02\ \mu_B$ ($R = Er$) to $0.08\ \mu_B$ ($R = Gd$) for $2c$ sites and from $0.01\ \mu_B$ (Er) to $0.04\ \mu_B$ (Gd) for $6i$ sites. These moments are antiparallely oriented to those of rare-earths.

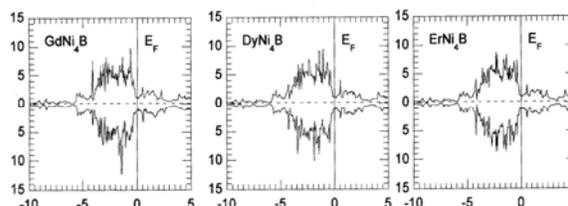


Fig. 2. Total densities of states, for RNi_4B ($R=Gd, Dy, Er$).

The $Ni(2c)$ magnetic contributions determined from band structure calculations are higher than those of $Ni(6i)$,

because of different local environments. The effective Ni moments are nearly constant and close to 1.8 μ_B /atom.

The Ni 2p_{1/2} and Ni 2p_{3/2} core level lines are plotted in Fig. 3. The values of the binding energies in RNi₄B compounds and pure Ni are listed in Table 1. The line positions are not modified in compounds as compared to pure Ni.

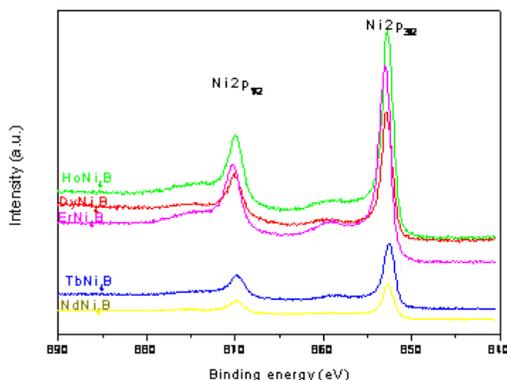


Fig. 3. Ni 2p_{1/2} and Ni 2p_{3/2} core levels.

Table 1. The binding energies of Ni 2p in Ni and RNi₄B compounds.

Compound	Ni 2p _{3/2}	Ni 2p _{1/2}
Ni	852.7 ± 0.1	869.97 ± 0.1
NdNi ₄ B	852.306 ± 0.1	869.4 ± 0.1
TbNi ₄ B	852.17 ± 0.1	869.37 ± 0.1
DyNi ₄ B	852.85 ± 0.1	870.1 ± 0.1
HoNi ₄ B	852.59 ± 0.1	869.83 ± 0.1
ErNi ₄ B	853.02 ± 0.1	870.48 ± 0.1

The Ni 6 eV satellite line appears in the RNi₄B compounds, with R = Tb, Dy, Ho, Er, at the same binding energy as in pure Ni (858 ± 0.1 eV). The line intensity is lower than in case of pure Ni, suggesting that the number of holes is diminished as compared to Ni 3d band. This is in agreement with the smaller effective Ni moment determined in the paramagnetic range (1.8 μ_B /atom as compared to μ_{ef} (Ni²⁺) = 2.83 μ_B /atom) [9]. In the case of NdNi₄B compound, the Ni 6 eV line intensity cannot be unambiguously analysed, due to the superposition with the 4d lines of the rare earth.

The rare earth 4d core level spectra of the rare earths are given in Fig. 4. The small shifts that appear in RNi₄B compounds compared to pure metals were attributed to minor changes in shielding which result from modification of electron concentration [10].

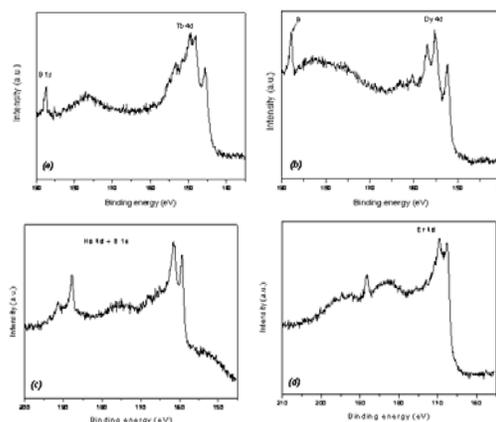


Fig. 4. Core level lines of RNi₄B compounds with R = Tb (a), Dy (b), Ho (c) and Er (d).

The Nd 3d core level lines and 4d ones of R = Dy, Ho, Tb, Er were decomposed in the main and satellite lines as in pure metals [11], and are given in Table 2. The binding energies show only small shifts as compared to the values determined in pure metals, indicating a similar electronic configuration. In the case of NdNi₄B compound, the 3d_{3/2} and 3d_{5/2} lines of Nd were identified at 1004.45 eV and 980.68 eV binding energy, respectively, little shifted as compared to pure Nd. This may correspond to a contamination with Nd₂O₃ of the sample surface. Also, the spin-orbit splitting of the 3d lines is higher than in case of pure metal (23.77 eV vs 22.6 eV).

Table 2. Binding energie (BE)s of rare earths in pure elements and in RNi₄B compounds.

Element	RE lines	BE in pure element (eV)	BE in compound (eV)
Nd	3d _{3/2}	1003	1004.45 ± 0.2
	3d _{5/2}	980.8	980.68 ± 0.2
Tb	4d	145.95	145.49 ± 0.2
		147.98	148.61 ± 0.2
		150.55	149.73 ± 0.2
		153.91	153.05 ± 0.2
Dy	4d	152.4	152.3 ± 0.2
		153.31	153.6 ± 0.2
		155.12	154.99 ± 0.2
		156.91	156.95 ± 0.2
Ho	4d	159.59	159.37 ± 0.2
		161.83	161.41 ± 0.2
Er	4d	167.3	167.67 ± 0.2
		169.3	169.63 ± 0.2

The valence bands spectra of RNi₄B compounds with R = Tb(a), Ho(b), Er(c) and Nd(d) are given in Fig. 5.

